Synthetic Methods

DOI: 10.1002/ange.201310487

Asymmetric Vinylogous Diels-Alder Reactions Catalyzed by a Chiral Phosphoric Acid**

Xu Tian, Nora Hofmann, and Paolo Melchiorre*

Abstract: An unprecedented way to extend the synthetic utility of the Diels—Alder reaction to include a vinylogous reactivity space is described. A commercially available chiral phosphoric acid catalyst effectively activates cyclic 2,4-dienones towards a vinylogous [4+2] cycloaddition with 2-vinylindoles, which leads to stereochemically dense tetrahydrocarbazoles. The reaction proceeds with a high level of remote stereocontrol and exclusive chemoselectivity for the more distant double bond of the dienone.

The Diels-Alder (DA) reaction is among the most powerful methods for the rapid and predictable construction of stereochemically dense cyclohexenyl rings.^[1] Research that is aimed at further expanding the synthetic potential of the DA approach is still fascinating the chemical community. Avenues for new progress have mainly been provided by the emergence of asymmetric catalytic variants^[2] and the desire to identify unprecedented diene-dienophile combinations. In contrast, the use of DA chemistry to stereoselectively shape the reactivity space that is remote from the catalyst point of action has remained largely underdeveloped. [3] This is probably due to the inherent difficulties of achieving remote stereoinduction, as a close spatial contact between the chiral catalyst and the reaction site is generally required. [4] Herein, we describe an effective method to carry out a catalytic vinylogous DA reaction with high control over the remote stereochemistry. This transformation relies upon the LUMOlowering activation of the more distant double bond of $\alpha,\beta,\gamma,\delta$ -unsaturated cyclic ketones 1 (Figure 1). Specifically, we used chiral Brønsted acid catalysis^[5] to increase the tendency of the distant γ,δ -olefinic moiety in 1 to react as a chiral dienophile, an activation principle that has never served to induce vinylogous reactivity to date. The reaction

[*] Prof. Dr. P. Melchiorre
ICREA—Institució Catalana de Recerca i Estudis Avançats
Passeig Lluís Companys 23, 08010 Barcelona (Spain)
X. Tian, Dr. N. Hofmann, Prof. Dr. P. Melchiorre
ICIQ—Institute of Chemical Research of Catalonia
Avenida Països Catalans 16, 43007 Tarragona (Spain)
E-mail: pmelchiorre@iciq.es
Homepage: http://www.iciq.es/portal/862/default.aspx

[**] Research support from the Institute of Chemical Research of Catalonia (ICIQ) Foundation and the European Research Council (ERC Starting Grant 278541 "ORGA-NAUT" to P.M.) is gratefully acknowledged. N.H. is grateful to the German Academic Exchange Service (DAAD) for a postdoctoral fellowship. We thank E. Escudero-Adán (X-ray Diffraction Unit, ICIQ) for determining the structure of compound 3 e.



Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/anie.201310487.

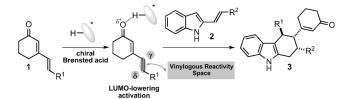


Figure 1. Design of a Brønsted acid catalyzed asymmetric vinylogous Diels—Alder reaction.

with 2-vinylindoles **2**, which can readily participate in [4+2] cycloaddition processes as electron-rich dienes, ^[6] allowed the direct synthesis of stereochemically dense tetrahydrocarbazoles **3**^[7] with very high regio-, diastereo-, and enantioselectivity.

We began our investigations by examining the reaction between cyclic 2,4-dienone 1a and styryl-1H-indole 2a (Table 1). The choice of substrates was based on our previous experiences with vinylogous reactivity. [8] We recently established that cinchona-alkaloid-based primary amines of type $A^{[9]}$ can condense with β -substituted cyclic dienones 1, which facilitates the formation of an extended iminium ion intermediate while lowering the LUMO energy level of the distant unsaturated π system. The resulting activated vinylogous iminium ion activation allowed for δ -site-selective and enantioselective nucleophilic 1,6-additions^[8a] and vinylogous cascade transformations to be realized. [8b] As the same activation principle underlies the Diels-Alder chemistry, we tested the ability of quinidine derivative A to catalyze this model reaction in an asymmetric vinylogous fashion (entries 1 and 2). Unfortunately, this catalytic system was ineffective, and provided the desired product 3a only with poor yield and stereoinduction. However, the vinylogous DA reaction is amenable not only to aminocatalysis, but also to a different activation mode. We found that a Brønsted acid[10] could promote the reaction by effectively activating the carbonyl moiety of dienone **1a**.^[5] An array of chiral 1,1'-bi-2-naphthol (BINOL)-derived phosphoric acids^[11] were screened as catalysts for the model reaction. Commercially available 4d was the best catalyst, as its use led to the formation of product 3a with complete control over the relative configuration and 94% ee (entry 6).[12] The more acidic[13] chiral phosphoramide $4e^{[10a]}$ induced an appreciable level of stereo- and regioselectivity, but could not parallel the efficacy of 4d (entry 7). A control experiment confirmed that the Brønsted acid catalyst is needed for the DA reaction to occur (entry 8), while protection of the N-H indole moiety in 2 greatly decreased the stereoselectivity of the process (entry 9). Mechanistically, the last result is consistent with a specific



Table 1: Explorative studies.[a]

Entry	Catalyst	R	Conv. ^[b] [%]	d.r. ^[c]	ee ^[d] [%]
1 ^[e]	Α	Н	< 5	n.d.	n.d.
2 ^[e]	Α	Me	55	1:1	27
3	4a	Н	20	> 20:1	77
4	4 b	Н	40	> 20:1	79
5	4 c	Н	< 5	n.d.	n.d.
6	4 d	Н	74	> 20:1	94
7	4e	Н	67	> 20:1	81
8	_	Н	< 5	n.d.	n.d.
9	4 d	Me	55	2:1	5
10 ^[f]	4 d	Н	88 ^[g]	> 20:1	94

[a] Reactions performed on a 0.05 mmol scale for 16 h with a catalyst (10 mol%), 2a or 2a' (1.2 equiv), and $[1a]_0 = 0.25$ M in PhMe. [b] Conversion of 1a determined by 1 H NMR analysis of the crude mixture using 1,3,5-trimethoxybenzene as the internal standard. [c] d.r. determined by 1 H NMR analysis of the crude reaction mixture. [d] ee determined by HPLC analysis on a chiral stationary phase. [e] Performed at 60 °C with A (20 mol%) and TFA (30 mol%) as the acid co-catalyst. [f] 4d (5 mol%), 2a (1.5 equiv), $[1a]_0 = 0.5$ M, 48 h. [g] Yield of isolated 3a n.d. = not determined, 16 Tifluoromethanesulfonyl, 16 TFA = trifluoroacetic acid.

and crucial interaction between substrate **2** and the catalyst, which productively adds to the acid-induced activation of the enone (see below for further discussions).

During a second round of optimization it was found that the use of 5 mol% of phosphoric acid 4d, a slight excess of 2a (1.5 equiv), and stirring in toluene at 40 °C for 48 hours provided adduct 3a with excellent conversion and selectivity (entry 10; 88%, >20:1 d.r., 94% ee). These reaction conditions were thus chosen for evaluating the synthetic potential and the generality of the vinylogous DA reaction. This method is synthetically useful, as a slightly higher efficiency was observed when running the model reaction on a 1 mmol scale (Table 2, entry 1).

A wide range of substituents at the δ -position of the dienone are compatible with the catalytic system. Different substitution patterns at the aromatic moiety were well tolerated, regardless of the electronic properties of the substituents, as the corresponding adducts **3** were obtained in good yields and with enantioselectivities of $\geq 90\%$ ee (entries 1–7). The synthesis of furyl derivative **3h** (entry 8) confirmed that a heteroaryl framework can be included in the

Table 2: Scope of the vinylogous DA reaction: variation of the cyclic 2,4-dienone. $^{[a]}$

Entry	R	n	3	Yield ^[b] [%]	ee ^[c] [%]
1 ^[d]	Ph	1	3 a	86	97
2	4-F-C ₆ H ₄	1	3 b	70	92
3	4-Cl-C ₆ H ₄	1	3 c	84	92
4	3-Cl-C ₆ H ₄	1	3 d	78	92
5 ^[e]	2-Br-C ₆ H ₄	1	3 e	80	93
6	4-Me-C ₆ H ₄	1	3 f	86	92
7 ^[f]	4-MeO-C ₆ H ₄	1	3 g	63	90
8	3-furyl	1	3 h	78	88
9	Me	1	3 i	85	89
10	Ph	0	3 j	60	91

[a] Reactions performed in air, without any precautions to exclude moisture, on a 0.1 mmol scale with $\bf 2a$ (1.5 equiv). For all of the reactions, a d.r. of > 20:1 was inferred by 1 H NMR analysis of the crude reaction mixture. [b] Yield of isolated $\bf 3$ after purification by column chromatography. [c] Determined by HPLC analysis on a chiral stationary phase. [d] 1 mmol scale. [e] $\bf 4d$ (10 mol%). [f] 72 h.

final product. Remarkably, this transformation can be extended to include substrates that bear an aliphatic substituent (R = Me, entry 9). Furthermore, a five-membered cyclic scaffold was a competent dienophile of this vinylogous DA reaction, providing the corresponding cycloadduct 3j with high stereocontrol and a slightly decreased chemical yield (entry 10). As a limitation of this system, acyclic 2,4dienones did not react at all under the optimized reaction conditions. From a synthetic standpoint, it is worth mentioning that for all of the reactions described in Table 2, products 3 were selectively formed as single diastereomers (complete control over the relative stereochemistry). Crystals that were obtained for compound 3e (entry 5) were suitable for X-ray crystallographic analysis. Thus, the stereochemical outcome of the Diels-Alder reaction as well as the absolute configurations of the three stereogenic centers could be established.[14]

We then evaluated the generality of this transformation by varying the structure of 2-vinylindole **2**, which is the diene component of the vinylogous DA reaction (Table 3). In general, different substitution patterns on the aryl ring that is attached to the exocyclic double bond ($R^1 = \text{aryl}$) were well tolerated (entries 1–5), although electron-withdrawing groups required the use of 10 mol% of catalyst for synthetically useful chemical yields (entries 1–3). The same behavior was observed for electron-withdrawing or -donating groups at the 5-position of the indole core (entries 6–8), the latter being more reactive towards the vinylogous cycloaddition pathway.

Importantly, the exocyclic double bond could also bear an aliphatic substituent ($R^1 = Me$; entry 9). In this case, a slightly modified procedure was needed: The diene was added portionwise to prevent a tandem homo-Diels-Alder cyclization/aromatization reaction of $2^{[15]}$ This experiment afforded

 $\begin{tabular}{ll} \textbf{Table 3:} & Scope of the vinylogous DA reaction: variation of the 2-alkenylindole. \end{tabular}$

Entry	R ¹	R ²	3	Yield ^[b] [%]	ee ^[c] [%]
1 ^[d,e]	4-Cl-C ₆ H₄	Н	3 k	64	93
$2^{[d,e]}$	4-Br-C ₆ H ₄	Н	31	53	92
3 ^[d]	2-Br-C ₆ H ₄	Н	3 m	95	90
4	4-Me-C ₆ H ₄	Н	3 n	74	93
5	$3,5-(MeO)_2-C_6H_3$	Н	3 o	94	95
6 ^[d,e]	Ph	Cl	3 p	80	92
7	Ph	Me	3 q	70	93
8 ^[e]	Ph	MeO	3 r	59	92
9 ^[d,f]	Me	Н	3 s	61	99

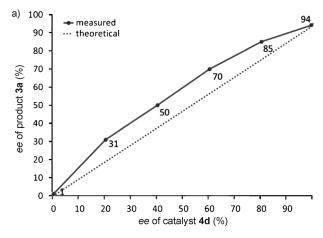
[a] Reactions performed on a 0.1 mmol scale using $\bf 2$ (1.5 equiv). For all of the reactions, a d.r. of > 20:1 was inferred by 1H NMR analysis of the crude reaction mixture. [b] Yield of isolated $\bf 3$ after purification by column chromatography. [c] Determined by HPLC analysis on a chiral stationary phase. [d] $\bf 4d$ (10 mol%). [e] 72 h. [f] Diene $\bf 2$ was added portionwise to the reaction mixture; using the standard procedure, $\bf 3s$ was isolated in only 20% yield.

the corresponding adduct **3s** with an excellent enantiomeric excess (99 % *ee*, > 20:1 d.r.).

We then focused on the mechanism of the transformation. As alluded to above, the decrease in stereoselectivity that is observed when the N-H group of diene 2 is protected (Table 1, entry 9) could suggest the requirement for a key interaction with the Brønsted basic (P=O) moiety of the catalyst 4d, alongside a concomitant Brønsted acid activation of the carbonyl group of dienophile 1 (P-OH···O=C). We have performed a nonlinear effect (NLE) study[16] of the model reaction to examine whether the optical purity of catalyst 4d directly correlated with the optical purity of product 3a.[17] A significant and positive NLE was observed, as shown by the convexity of the curve in Figure 2a, which indicates that more than one molecule of the chiral acid 4d is likely to be involved in the transition state of the enantiodifferentiating step. A mechanistic model that accounts for these experimental observations is proposed in Figure 2b.^[18]

One peculiar feature of this catalytic asymmetric vinylogous DA reaction is that it provides direct and rapid access to tetrahydrocarbazoles $\bf 3$ while preserving an α,β -unsaturated system. This functional group could be used as a suitable chemical handle for easily increasing the molecular and stereochemical complexity of the products by simple follow-up transformations. To illustrate this synthetic potential, the cyclohexenone moiety of $\bf 3a$ was stereoselectively converted into epoxyketone $\bf 5$ by means of a protection/epoxidation sequence (Scheme 1a). Furthermore, we could perform a palladium-catalyzed Suzuki–Miyaura cross-coupling with tetrahydrocarbazole $\bf 3m$ without requiring an additional protection step, which led to product $\bf 6$ (Scheme 1b).

In summary, we have developed a highly stereo- and regioselective vinylogous Diels-Alder reaction that is cata-



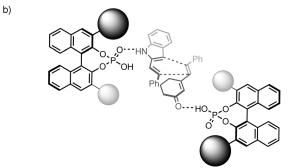


Figure 2. a) NLE study. b) Proposed stereochemical model.

Scheme 1. Synthetic manipulations of the DA adducts. Boc = tertbutoxycarbonyl, DMAP=4-dimethylaminopyridine, DME=1,2-dimethoxyethane, dppf=diphenylphosphinoferrocene.

lyzed by a commercially available chiral phosphoric acid. A range of structurally diverse complex tetrahydrocarbazoles could be synthesized with high chemical yield and excellent stereoselectivity. It was revealed that cyclic dienones can be activated towards a cycloaddition pathway by means of phosphoric acid catalysis. Furthermore, it was demonstrated that the synthetic utility of the Diels-Alder reaction can be extended to include a vinylogous reactivity space.



Received: December 3, 2013 Revised: December 19, 2013 Published online: February 6, 2014

Keywords: asymmetric catalysis · Brønsted acids · Diels-Alder reactions · organocatalysis · synthetic methods

- [1] a) O. Diels, K. Alder, Justus Liebigs Ann. Chem. 1926, 450, 237; b) E. J. Corey, Angew. Chem. 2002, 114, 1724; Angew. Chem. Int. Ed. 2002, 41, 1650; c) K. C. Nicolaou, S. A. Snyder, T. Montagnon, G. Vassilikogiannakis, Angew. Chem. 2002, 114, 1742; Angew. Chem. Int. Ed. 2002, 41, 1668.
- [2] Y. Hayashi, "Catalytic asymmetric Diels-Alder reactions" in Cycloaddition reactions in organic synthesis (Eds.: S. Kobayashi, K. A. Jørgensen), Wiley-VCH, Weinheim, 2001.
- [3] A viable strategy towards catalytic enantioselective vinylogous DA reactions has recently been identified; it involves the transient formation of chiral trienamine intermediates that can readily participate in DA processes as activated chiral dienes; see: a) Z. J. Jia, H. Jiang, J.-L. Li, B. Gschwend, Q.-Z. Li, X. Yin, J. Grouleff, Y.-C. Chen, K. A. Jørgensen, J. Am. Chem. Soc. 2011, 133, 5053; b) X.-F. Xiong, Q. Zhou, J. Gu, L. Dong, T.-Y. Liu, Y.-C. Chen, Angew. Chem. 2012, 124, 4477; Angew. Chem. Int. Ed. 2012, 51, 4401; c) Y. Liu, M. Nappi, E. Arceo, S. Vera, P. Melchiorre, J. Am. Chem. Soc. 2011, 133, 15212.
- [4] J. Clayden, Chem. Soc. Rev. 2009, 38, 817.
- [5] Sporadic examples have demonstrated the potential of chiral Brønsted acid catalysis for activating unsaturated ketones towards conjugate additions; see: a) H.-Y. Tang, A.-D. Lu, Z.-H. Zhou, G.-F. Zhao, L.-N. He, C.-C. Tang, Eur. J. Org. Chem. 2008, 1406; b) M. Rueping, B. J. Nachtsheim, S. A. Moreth, M. Bolte, Angew. Chem. 2008, 120, 603; Angew. Chem. Int. Ed. 2008, 47, 593; c) T. Akiyama, T. Katoh, K. Mori, Angew. Chem. 2009, 121, 4290; Angew. Chem. Int. Ed. 2009, 48, 4226; d) Z. Zhang, J. C. Antilla, Angew. Chem. 2012, 124, 11948; Angew. Chem. Int. Ed. 2012, 51, 11778.
- [6] For Diels-Alder reactions with 2-vinylindole derivatives, see: a) S. B. Jones, B. Simmons, D. W. C. MacMillan, J. Am. Chem. Soc. 2009, 131, 13606; b) C. Zheng, Y. Lu, J. Zhang, X. Chen, Z. Chai, W. Ma, G. Zhao, Chem. Eur. J. 2010, 16, 5853; c) X.-F. Wang, J.-R. Chen, Y.-J. Cao, H.-G. Cheng, W.-J. Xiao, Org. Lett. 2010, 12, 1140; for an early example of the use of 3-vinylindoles, see: d) C. Gioia, A. Hauville, L. Bernardi, F. Fini, A. Ricci, Angew. Chem. 2008, 120, 9376; Angew. Chem. Int. Ed. 2008, 47, 9236.
- [7] Tetrahydrocarbazoles have been recognized as important structural motifs and are found in a variety of natural products and pharmacologically active compounds; see: a) J. E. Saxton, Nat. Prod. Rep. 1997, 14, 559; b) A. Nikitenko, D. Evrard, A. L. Sabb, R. L. Vogel, G. Stack, M. Young, M. Lin, B. L. Harrison, J. R. Potoski, Org. Process Res. Dev. 2008, 12, 76, and references therein.
- [8] a) X. Tian, Y. Liu, P. Melchiorre, Angew. Chem. 2012, 124, 6545; Angew. Chem. Int. Ed. 2012, 51, 6439; b) X. Tian, P. Melchiorre, Angew. Chem. 2013, 125, 5468; Angew. Chem. Int. Ed. 2013, 52, 5360.

- [9] P. Melchiorre, Angew. Chem. 2012, 124, 9886; Angew. Chem. Int. Ed. 2012, 51, 9748.
- [10] For Diels-Alder reactions of enones that are catalyzed by Brønsted acids, see: a) D. Nakashima, H. Yamamoto, J. Am. Chem. Soc. 2006, 128, 9626; b) D. Nakashima, H. Yamamoto, Org. Lett. 2005, 7, 1251; c) T. Schuster, M. Bauch, G. Durner, M. W. Göbel, Org. Lett. 2000, 2, 179; for an asymmetric catalytic version that is based on iminium ion activation, see: d) A. B. Northrup, D. W. C. MacMillan, J. Am. Chem. Soc. 2002, 124, 2458.
- [11] a) T. Akiyama, Chem. Rev. 2007, 107, 5744; b) M. Terada, Synthesis 2010, 1929; for pioneering studies on chiral BINOLderived phosphoric acid catalysis, see: c) T. Akiyama, J. Itoh, K. Yokota, K. Fuchibe, Angew. Chem. 2004, 116, 1592; Angew. Chem. Int. Ed. 2004, 43, 1566; d) D. Uraguchi, M. Terada, J. Am. Chem. Soc. 2004, 126, 5356.
- [12] A reactivity- and selectivity-enhancing effect of Ca²⁺ ions in combination with phosphoric acid catalysts has been reported; see: M. Hatano, K. Moriyama, T. Maki, K. Ishihara, Angew. Chem. 2010, 122, 3911; Angew. Chem. Int. Ed. 2010, 49, 3823. We performed a control experiment using a sample of commercially available catalyst 4d that had been further purified by a careful extraction with HCl. This experiment provided a similar reactivity and stereoselectivity (3a obtained in 88% yield, 88% ee), indicating that metals are not involved in the reported reaction.
- [13] For information on the acidity of chiral phosphoric acids, see: a) P. Christ, A. G. Lindsay, S. S. Vormittag, J.-M. Neudörfl, A. Berkessel, A. C. O'Donoghue, Chem. Eur. J. 2011, 17, 8524; b) K. Kaupmees, N. Tolstoluzhsky, S. Raja, M. Rueping, I. Leito, Angew. Chem. 2013, 125, 11783; Angew. Chem. Int. Ed. 2013, 52, 11569.
- [14] CCDC 973586 contains the supplementary crystallographic data for this paper (3e). These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www. ccdc.cam.ac.uk/data_request/cif.
- [15] This pathway can be catalyzed by a strong Brønsted acid as previously reported; see: C. B. Chen, X. F. Wang, Y. J. Cao, H. G. Cheng, W.-J. Xiao, J. Org. Chem. 2009, 74, 3532.
- [16] For a review, see: a) T. Satyanarayana, S. Abraham, H. B. Kagan, Angew. Chem. 2009, 121, 464; Angew. Chem. Int. Ed. 2009, 48, 456; for a recent report on a negative NLE associated with chiral phosphoric acid catalysis, see: b) C. K. De, F. Pesciaioli, B. List, Angew. Chem. 2013, 125, 9463; Angew. Chem. Int. Ed. 2013, 52, 9293
- [17] Both the optically pure phosphoric acid 4d and its racemic mixture proved completely soluble under the reaction conditions (toluene, $[4d]_0 = 0.025 \,\mathrm{M}$, stirring at 40 °C for 16 h). A positive NLE was also observed when performing the experiments in chloroform. For a pertinent discussion, see: N. Li, X.-H. Chen, S.-M. Zhou, S.-W. Luo, J. Song, L. Ren, L.-Z. Gong, Angew. Chem. 2010, 122, 6522; Angew. Chem. Int. Ed. 2010, 49, 6378.
- Further investigations are needed to fully clarify the reaction mechanism. Specifically, we are making use of a combination of experimental and theoretical mechanistic studies to ascertain whether a concerted or a stepwise mechanism is operative in our model reaction.